# COMPUTER SIMULATION OF EQUILIBRIUM MULTICOMPONENT MULTISTAGE SEPARATION PROCESSES

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In Partial Fulfilment of the Requirements
for the Degree of
MASTER OF TECHNOLOGY

by SAHIDUL ISLAM

to the

DEPARTMENT OF CHEMICAL ENGINEERING
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# CERTIFICATE

This is to certify that the work presented in this thesis entitled, " COMPUTER SIMULATION OF EQUILIBRIUM MULTICOMPONENT MULTISTAGE SEPARATION PROCESSES " has been earried out by Mr. Sahidul Islam under my supervision and the same has not been submitted elsewhere for a degree.

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I would be extremely happy to convey my sincere thanks to all my friends and seniors in different fields who helped me directly and indirectly for the completion of this thesis. Special thanks are due to Dr. Ratan Mohan, Dr. A.K. Verma, Mr. Raghu Raman, Mr. Mrityunjoy Chakroborty, Mr. Binoy Bhusan Kandir, Dr. Sujit Dutta, Mr. Susil Mandal, Mr. Kausik Banerjee, Mr. Tapas Mandal, Mr. S.R. Diksitulu, Mr. Alok Pandit, Mr. VLN Murthy, Mr. Saibal Banerjee, Mr. M.M. Beg, Dr. P.K. Bhatt and Miss Arti Gupta.

I do not know how I will express my deep appreciation to our immortal beloved Newton. As a child I am trying to understand and to apply one of his principle, which is nothing compare to his major contributions, but is the heart of this thesis.

Finally what I feel that help and inspiration is as if solely from \* Marrya who boosted me and thereby helping me to think the problem, has been inducing and driving me to my ultimate long desire.

Sahidul Islam

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	for nonideal system.	

# ABSTRACT

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Computer Simulation of Equilibrium Title of the Thesis

Multicomponent Multistage Separation

Processes

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The modified Thomas algorithm is employed to obtain the correction vector  $\Delta \overline{X}$  in the widely used Naphthali-Sandholm method of solving separation processes problems. An efficient algorithm for obtaining the correction AX is proposed in which the sparsity of the submatrices of the Jacobian has been expoited in the matrix multiplication and inversion. The operation count for the proposed algorithm and the standard matrix multiplication and inversion has been presented using two bench-mark problems. It has been shown that the use of the proposed algorithm results in considerable saving in the CPU time with increase in the number of components and stages. The proposed algorithm can be used for the extensions and variants of Naphthali-Sandholm methods.

#### CHAPTER 1

# INTRODUCTION

The classical techniques of separation of multicomponent mixtures like distillation, absorption, extraction, etc. are widely used in chemical industries. The design or simulation of these (stage-wise) separation processes involves the solution of the material and energy balance equations and equilibrium relations for each stage, and requires an enormous amount of computational effort. Before the widespread use of computers, short-out methods were employed for the design though these are generally inadequate for systems other than the ones for which the equilibrium relations are linear. But with accessability to powerful computers, the rigorous methods got impetus and several methods have been proposed.

The rigorous methods of design or simulation involve two major steps; namely formulation of basic equations and their numerical methods of solution. The methods of formulation of the basic equations, in turn, can be classified as componentwise grouping of variables and stage-wise grouping of variables. The earlier methods of solution like the B-P method, the sumrate method and the relaxation technique can be visualized as the direct substitution methods. Recently, these methods of

solution have given way to the more efficient Newton-Raphson method and its variants.

Under the component-wise grouping of variables, the basic equations can be obtained as

$$\ddot{C}_{i} X_{i} = - T_{i} \quad \text{for } i = 1, 2 \dots C$$
 (1)

for the steady state conditions; where  $C_i$  is the coefficient matrix involving vapor and liquid flow rates and the k-values,  $\overline{X}_i$  could be either liquid or vapor mole fractions or component flow rates,  $\overline{f}_i$  are the feed component flow rates and i the components. In the relaxation technique (also known as False-Transient method) the basic equations are cast as

where k is the iteration number.

More generally, the basic equations may be represented as

$$\bar{A}_{\underline{i}} X_{\underline{i}} = \bar{b}_{\underline{i}} \qquad \underline{i} = 1, 2 \dots . C$$
 (3)

The square matrix has the tridiagonal structure and  $\overline{X}_i$  can be found using the well known Thomas algorithm.

Equation (3) together with the enthalpy balance equations around the stages can be solved by the direct

substitution method for any assumed liquid and vapor flow rates and temperatures. The B-P method, the sum-rate method and the modified relaxation technique fall under this category.

Since the year 1965, the direct substitution methods gave way to the Newton-Raphson method(or its modified versions) of solution. In these methods, the material balance discrepancy functions,

$$M_{n} = \sum_{i=1}^{C} y_{i,n} - \sum_{i=1}^{C} x_{i,n} \quad n = 1, 2... N$$
 (4)

and the enthalpy discrepancy functions around each stage are expanded in the Taylor series. Assuming the second and higher order terms to be negligible, the equations are rearranged to obtain

$$\bar{J} \Delta \bar{X} = \bar{b} \tag{5}$$

where J is the Jacobian matrix of 2Nx2N.

Several convergence schemes and the methods of obtaining the correction vector have been proposed. These have been discussed by Holland (2).

Naphthali (4) and later Naphthali and Sandholm (5) showed that the convergence characteristics are better if the stage-wise grouping of variables together with the Newton-Raphson method are employed. For each of the stages the

variables are the component liquid flow rates, temperature and the component vapor flow rates (i.e. l<sub>i</sub>,T,v<sub>i</sub>). The component and the enthalpy discrepency functions are formulated to yield

$$O = (\overline{X}) \overline{Y}$$

Applying the Newton-Raphson technique wa get

$$J_{AX} = -\overline{F}$$

The Jacobian is of the order (2C+1) N x(2C+1)N and it has the block tridiagonal form. Several methods proposed for solving the separation problems deal with the techniques of obtaining the correction vector, Stadther [6, 7, 8].

It has been generally accepted that the Naphthali-Sandholm method has better convergence characteristics. However, it is likely to diverge if the guessed component liquid and vapor flow rates and temperatures are far from the correct values. To over come this problem, Ketchum [3] proposed the fusion of the relaxation technique and the Naphthali-Sandholm method and demonstrated its suitablity for even interlinked columns. Later Hofeling and Seader [1] have demonstrated as to how the modified Thomas algorithm can be used for the interlinked columns. Stadther [6,7,8] has compared the

convergence characteristics of the modified Thomas algorithm and the other methods of solution using the sparse matrix technique. He finds the Thomas algorithm as good as the other sparse matrix techniques.

From the literature, it appears that the sparsity of the submatrices (diagonal and its adjacent submatrix) has not been exploited. The objective of the present work is to propose a more efficient method of solution taking advantage of the sparsity of the submatrices.

In Chapter 2, the method of formulation of the basic equations and the method of solution is presented. The results and discussion are presented in Chapter 3. Next conclusions is presented.

#### CHAPTER 2

# BASIC EQUATIONS AND METHOD OF SOLUTION

In this Chapter, the strategy of formulating the basic equations involving material and enthalpy balances, equilibrium and tray efficiency relations of multistage multicomponent separation processes, and the method of obtaining solution is presented.

A model of a general stage, along with the notation employed herein, is depicted in figure 1. The various discrepancy functions can be formulated as follows:

For Material balance:

$$M_{i,n} = 1_{i,n-1} + f_{i,n-1} + f_{i,n} - (1+s)v_{i,n} - (1+s)l_{i,n}$$
for  $1 \le i \le C$  &  $1 \le n \le N$ 

where v<sub>i,n</sub> and l<sub>i,n</sub> are the flow rates of component i in liquid and vapor leaving the nth stage, and S<sub>n</sub>v<sub>n</sub> and s<sub>n</sub>l<sub>m</sub> are the vapor and liquid side streams drawn on the nth stage.

For Enthalpy balance:

$$E_{n} = \sum_{i=1}^{c} 1_{i,n-1}h_{i,n-1} + \sum_{i=1}^{c} v_{i,n+1} H_{i,n+1} + \sum_{i=1}^{c} f_{i,n}h_{F_{i,n}} - \sum_{i=1}^{c} (1+S_{n})v_{i,n} H_{i,n} - \sum_{i>1} (1+S_{n}) 1_{i,n}h_{i,n} + q_{n}....$$
 (2)

for  $1 \le i \le C$  &  $1 \le n \le N$ 

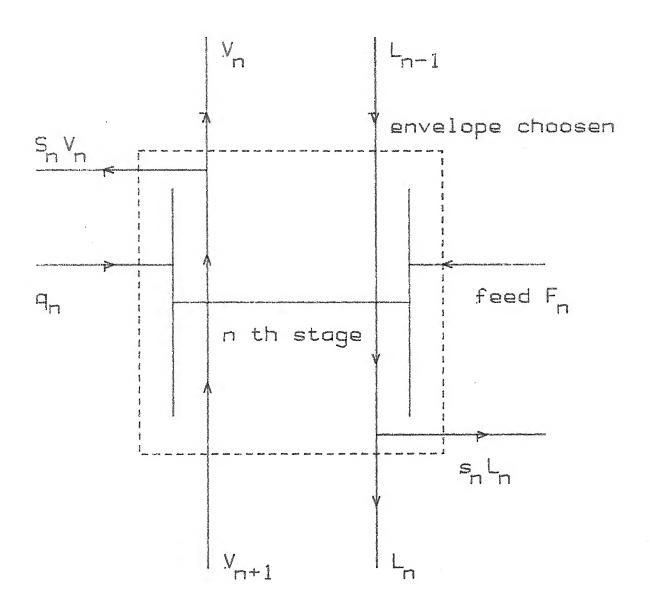


Fig. 1: A typical contacting stage

Equilibrium relations together with the Murphree efficiency:

$$O_{i,n} = \frac{\gamma_{n} k_{i,n} l_{i,n}}{L_{n}} - \frac{v_{i,n}}{v_{n}} + (1 - \gamma_{n}) \frac{v_{i,n+1}}{v_{n+1}} \cdots$$
 (3)

for  $1 \le i, j \le C$  &  $1 \le n \le N$ 

where 
$$\eta_n = \frac{y_{i,n} - y_{i,n+1}}{k_{i,n} x_{i,n} - y_{i,n+1}}$$
 (3a)

Thus, there are N(2C+1) set of nonlinear equations and they have, to be solved to obtain the N(2C+1) unknown variables, named i,n, vi,n and  $T_n$ .

The set of equations may be compactly written as

$$\overline{F} = \overline{F} (\overline{X}) = \overline{O}$$
 (4)

where

$$\overline{F} = \left[\overline{F}_1, \overline{F}_2 \dots \overline{F}_n \dots \overline{F}_N\right]^T, \qquad (5)$$

$$F_n = [M_{1,n}, M_{2,n}, M_{C,n}, O_{1,n}, O_{2,n}, M_{C,n}]^T$$
(6)

$$\overline{\mathbf{X}} = \left[ \overline{\mathbf{X}}_{1}, \overline{\mathbf{X}}_{2}, \dots, \overline{\mathbf{X}}_{m} \right]^{\mathrm{T}}$$
 (7)

and 
$$\overline{X}_{n} = \begin{bmatrix} 1_{1,n}, 1_{2,n}, \dots & 1_{C,n}, v_{1,n}, v_{2,n}, \dots & v_{C,n}, v_{n} \end{bmatrix}^{T}$$
 (8)

It may be pointed out that the ordering of the variables is slightly different from the one proposed by Naphthali and

Sandholm 1971 . The reasons for the order chosen here is explained later.

Employing the Newton-Raphson technique, from Equation(4) we can obtain

$$\Delta \overline{X} = -\frac{\sqrt{F}}{\sqrt{X}}$$

$$= -\frac{1}{J} \cdot \overline{F}$$

where J is the Jacobian matrix.

The Jacobian has the block tridiagonal structure and can be represented as

where

$$\bar{\bar{A}}_n = \frac{\partial \bar{\bar{Y}}_n}{\partial \bar{\bar{X}}_{n-1}}, \quad \bar{\bar{B}}_n = \frac{\partial \bar{\bar{Y}}_n}{\partial \bar{\bar{X}}_n} \quad \text{and} \quad \bar{\bar{C}}_n = \frac{\partial \bar{\bar{Y}}_n}{\partial \bar{\bar{X}}_{n+1}}$$

and the rest of the elements are null matrices.

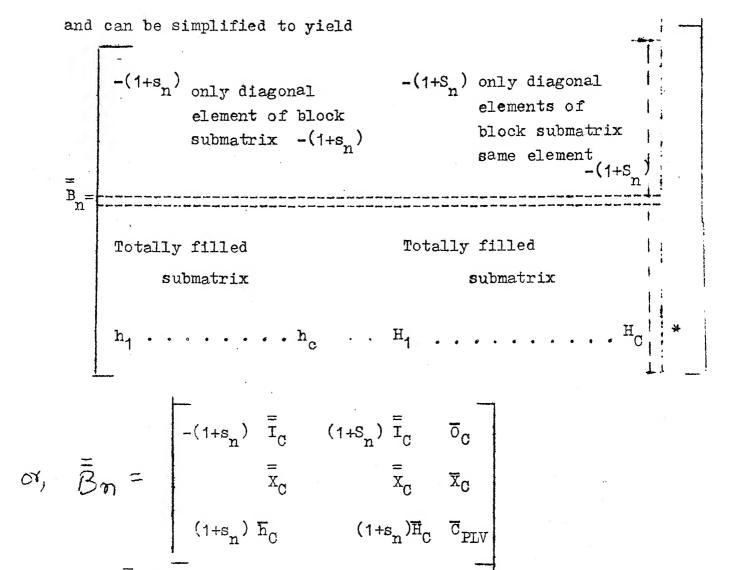
In expand form, the submatrix  $\bar{\bar{A}}_n$  is

$$\bar{A}_{n} = \begin{bmatrix} \overline{\partial}_{1,n}^{M}, & \overline{\partial}_{1,n}^{M}$$

where  $\overline{I}$ ,  $\overline{O}$ ,  $\overline{O}$  are identity matrix, null matrix and null vector respectively. Subscript C indicates the dimension of the matrix or vector.  $\overline{h}$  is the liquid enthalpy vector  $(h_i, i=1, C)$  and  $C_{PL}$  is heat capacity of  $L_{n-1}$ .

In expanded form, the submatrix  $\mathbf{E}_{n}$  is

Tu ex	changed totim, one adoma of the in	
	$\frac{\overline{\mathcal{D}^{M}_{1,n}}}{\overline{\mathcal{D}^{1}_{1,n}}}, \frac{\overline{\mathcal{D}^{M}_{1,n}}}{\overline{\mathcal{D}^{1}_{2,n}}} \cdots \frac{\overline{\mathcal{D}^{M}_{1,n}}}{\overline{\mathcal{D}^{1}_{C,n}}}, \frac{\overline{\mathcal{D}^{M}_{1,n}}}{\overline{\mathcal{D}^{V}_{1,n}}} \cdots$	$\frac{\partial M_{1,n}}{\partial v_{C,n}}, \frac{\partial M_{1,n}}{\partial T_n}$
	$\frac{\partial^{M_{2,n}}}{\partial^{l_{1,n}}}, \frac{\partial^{M_{2,n}}}{\partial^{l_{2,n}}} \cdots \frac{\partial^{M_{2,n}}}{\partial^{l_{C,n}}}, \frac{\partial^{M_{2,n}}}{\partial^{v_{1,n}}} \cdots$	$\frac{\partial M_{2,n}}{\partial v_{C,n}}$ , $\frac{\partial M_{2,n}}{\partial T_n}$
		••••
1	1 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	$\frac{\partial^{M_{C,n}}}{\partial \nabla_{C,n}}$ , $\frac{\partial^{M_{C,n}}}{\partial T_{m}}$
= B <sub>m</sub> =	$\frac{\partial O_{1,n}}{\partial I_{1,n}}$ , $\frac{\partial O_{1,n}}{\partial I_{2,n}}$ $\frac{\partial O_{1,n}}{\partial I_{C,n}}$ , $\frac{\partial O_{1,n}}{\partial V_{1,n}}$	$\cdot \frac{\partial^{0}_{1,n}}{\partial \mathbf{v}_{0,n}} \cdot \frac{\partial^{0}_{1,n}}{\partial \mathbf{r}_{n}}$
		• • • • • • • • • • • • • • • • • • • •
	$\frac{\partial o_{C,n}}{\partial l_{1,n}}$ , $\frac{\partial o_{C,n}}{\partial l_{2,n}}$ $\frac{\partial o_{C,n}}{\partial o_{C,n}}$ , $\frac{\partial o_{C,n}}{\partial v_{1,n}}$	$\frac{9 r^{C'u}}{9 o^{C'u}}, \frac{3 r^u}{9 o^{C'u}}$
	$\frac{\partial E_n}{\partial E_n}, \frac{\partial E_n}{\partial E_n}, \frac{\partial E_n}{\partial E_n}, \frac{\partial E_n}{\partial E_n}, \frac{\partial E_n}{\partial E_n}$	$\frac{\partial^{E}_{n}}{\partial^{v}_{C,n}}$ , $\frac{\partial^{E}_{m}}{\partial^{T}_{n}}$



where X, X and X denote the filled matrix, vector and nonzero element respectively and the subscript C the orders of the matrix or vector.

	$\frac{\partial^{M}_{1,n}}{\partial^{1}_{1,n+1}}, \frac{\partial^{M}_{1,n}}{\partial^{1}_{2,n+1}}$	$\frac{\partial^{\mathrm{M}_{1}}}{\partial^{\mathrm{M}_{2}}}$	$\frac{n}{n+1}$ , $\frac{\partial^{M}_{1,n}}{\partial^{V}_{1,n+1}}$	$\frac{\partial^{M_{1,n}}}{\partial^{V_{C,n+1}}}$	$\partial^{M_{1,n}}$
	$\frac{\partial^{\mathrm{M}}_{2,\mathrm{n}}}{\partial^{\mathrm{l}_{1,\mathrm{n+1}}}}, \frac{\partial^{\mathrm{M}}_{2,\mathrm{n}}}{\partial^{\mathrm{l}_{2,\mathrm{n+1}}}}$	$\frac{\partial^{M}_{2}}{\partial^{1}c},$	$\frac{n}{n+1}$ , $\frac{\partial^{M}_{2,n}}{\partial^{V}_{1,n+1}}$	$\frac{\partial^{M}_{2,n}}{\partial^{v}_{C,n+1}}$	$\frac{\partial^{M}_{2,n}}{\partial^{T}_{n+1}}$
	*****	• • • • • • • • • • • • • • • • • • •	• • • • • •	• • • • •	• • •
= C <sub>n</sub> =	$\frac{\partial^{M}_{C,n}}{\partial^{1}_{1,n+1}}, \frac{\partial^{M}_{C,n}}{\partial^{1}_{2,n+1}}$	$\frac{3_{1}^{C}}{9_{W}^{C}}$	$\frac{n}{n+1}$ , $\frac{\partial^{V_{C,n}}}{\partial^{W_{C,n}}}$	$\frac{2^{N_{C,n}}}{2^{N_{C,n+1}}}$	$\frac{3\pi^{n+1}}{2}$
	$\frac{\partial^{\circ}_{1,n}}{\partial^{\downarrow}_{1,n+1}}, \frac{\partial^{\circ}_{1,n}}{\partial^{\downarrow}_{1,n+1}}$	70 <sub>1</sub> ,	$\frac{n}{n+1}, \frac{\partial^0_{1,n}}{\partial v_{1,n+1}}$	$\frac{\partial \circ_{1,n}}{\partial v_{1,n+1}}$	$\frac{2^{\circ}_{1,n}}{2^{T_{n+1}}}$
					• • •
	$\frac{\partial O_{C,n}}{\partial I_{1,n+1}}, \frac{\partial O_{C,n}}{\partial I_{2,n}}$	δο <sub>C</sub>	$\frac{1}{\sqrt{n}}$ , $\frac{\sqrt{0_{C,n}}}{\sqrt{v_{1,n+1}}}$	$\frac{3^{\circ}_{0,n}}{3^{\circ}_{0,n+1}}$	$\partial_{C,n}$
	$\frac{\partial E_n}{\partial I_{1,n+1}}, \frac{\partial E_n}{\partial I_{2,n}}$	$\frac{\partial E_n}{\partial L_{C_1}}$	$\frac{\partial E_n}{\partial v_{1,n+1}}$	$\frac{\partial^{E_n}}{\partial^{v_{C,n+1}}}$	$\frac{\partial E_n}{\partial T_{n+1}}$
	1				

unit block submatrix null column vector Totally filled block sub matrix null block matrix and can be simplified to get

# THOMAS ALGORITHM

The correction vector  $\sqrt{X}$  can be found by the Thomas algorithm and is given below

Forward Substitution:

For the stage 1

Steps: 1 
$$P_1 \leftarrow (B_1)^{-1} C_1$$

$$: 2 \quad \overline{\mathbb{Q}}_1 \longleftarrow (\overline{\mathbb{B}}_1)^{-1} \ \overline{\mathbb{F}}_1$$

For the stages n from 2 to (N-1)

$$\vdots 3 \quad \stackrel{=}{\mathbb{P}}_{n} \leftarrow (\stackrel{=}{\mathbb{B}}_{n} - \stackrel{=}{\mathbb{A}}_{n} \stackrel{=}{\mathbb{P}}_{n-1})^{-1} \stackrel{=}{\mathbb{C}}_{n}$$

: 4 
$$\overline{\mathbb{Q}}_{n} \leftarrow (\overline{\mathbb{B}}_{n} - \overline{\mathbb{A}}_{n} \overline{\mathbb{P}}_{n-1})^{-1} (\overline{\mathbb{F}}_{n} - \overline{\mathbb{A}}_{n} \overline{\mathbb{Q}}_{n-1})$$

For the last stage N

$$: 5 \quad \overline{Q}_{N} \leftarrow (\overline{B}_{N} - \overline{A}_{N} \overline{P}_{N-1})^{-1} (\overline{F}_{N} - \overline{A}_{N} \overline{Q}_{N-1})$$

Backward substitution

: 6 
$$\triangle \overline{x}_N \leftarrow \overline{Q}_N$$

For the stages n from N-1 to 1

It can be seen that the algorithm involves several time taking matrix multiplications and inversions. Advantage of the sparcity and the structure of these matrices can be taken to minimise the machine operations, as described below.

In step 1 and 3, we encounter the matrix multiplications  $(\bar{B}_j)^{-1}\bar{C}_1$  and  $(\bar{B}_j-\bar{A}_j\bar{P}_j)^{-1}\bar{C}_j$ . The inverted matrices are totally filled while  $\bar{C}_1$  and  $\bar{C}_j$  are highly sparse. The structure of the resulting matrix is

Filled matrix
$$\begin{bmatrix}
\overline{c} & \overline{c} & \overline{c} & \overline{c} \\
\overline{c}$$

where X, X and X denote the filled matrix, vector and non-zero element respectively, X and X are the null and identity matrices, and the subcript X the order of the matrix or vector.

It can be seen that there is no need to compute the elements of the first C columns of the resulting matrix and we can save C (2C+1)<sup>2</sup> operations. Further, the next C columns can be found as

$$P_{j,C+i} = b_{j,i+C}^{I} + \sum_{K=C+I}^{2C+I} b_{j,k}^{I} C_{K,C+i} \text{ for } 1 \leq j \leq 2C+1$$

$$1 \leq i \leq C \qquad (3)$$

The elements of the last column are

$$P_{j,2C+1} = b_{j,2C+1}^{I} C_{2C+1,2C+1}$$
 (4)

Thus the operation count of this matrix multiplication is  $(C^2+1)(2C+1)$  instead of  $(2C+1)^3$  required for standard matrix multiplication.

The matrix multiplication  $\stackrel{=}{A}_j \stackrel{=}{P}_{j-1} (= \stackrel{=}{\alpha}_j)$  carried out taking the advantage of sparcity as given below.

Here, only the  $\overline{X}_C$ , X in the row need to be computed. The nonzero elements of  $\stackrel{=}{\swarrow}_n$  are:

Thus we need to perform only  $(C+1)^2$  scalar multiplication in stead of  $(2C+1)^3$ .

The product 
$$\overline{A}_n$$
  $\overline{F}_{n-1} = \overline{P}_{n-1}$  may be found as follows  $\overline{A}_0$   $\overline{$ 

where 
$$\beta_1 = \overline{F}_1$$
 for  $1 \leq J \leq C$  (8)

$$P_{2C+1} = \sum_{k=1}^{C} A_{2C+1,k} F_k + A_{2C+1,2C+1} F_{2C+1}$$
 (9)

only the last element need to be computed; thus only C+1 operations are required instead of  $(2C+1)^2$  operations for the standard matrix multiplication. In backward substitution, the multiplication  $P_{n+1}$  involves (2C+1)(C+1) since the first C columns are zero elements.

Inversion of  $(B_n - A_n P_{n-1})$  by partitioning.

Inversion by partitioning does not save the operations to be performed. But from the structure of the matrix  $(B_n - A_n P_{n-1})$  some saving in operations can be realised as shown below.

shown below.
$$(\overline{B}_{n} - \overline{A}_{n} \overline{P}_{n-1}) = \overline{b} = \begin{bmatrix} \overline{X}_{C} & \overline{X}_{C} & \overline{X}_{C} \\ \overline{X}_{C} & \overline{X}_{C} & \overline{X}_{C} \end{bmatrix}$$

$$\overline{X}_{C} \qquad \overline{X}_{C} \qquad \overline{X}_{C} \qquad X$$

$$= \begin{bmatrix} = \\ b_{11} & b_{12} \\ \overline{b}_{21} & b_{22} \end{bmatrix}$$

The inverse is given by

where 
$$\bar{D}_{11} = (\bar{b}_{11})^{-1} + (\bar{b}_{11})^{-1} \bar{b}_{12}$$
  $\bar{D}_{22}$   $\bar{D}_{21} = (\bar{b}_{11})^{-1} + (\bar{b}_{11})^{-1} \bar{b}_{12}$   $\bar{D}_{22} = (\bar{b}_{11})^{-1} \bar{b}_{12}$   $\bar{D}_{22} = (\bar{b}_{22} - \bar{b}_{21}(\bar{b}_{11})^{-1} \bar{b}_{12})^{-1}$ 

The inverse of b<sub>11</sub> can, in turn, be found by partitioning as given below.

It may be noted  $\bar{\mathbf{a}}_{11}$  is a diagonal matrix and its inversion can be obtained by simply taking the reciprocal of each of the diagonal elements. Thus to find the inverse of a matrix of 2C+1, we need only the standard inversion of matrix of order C and some matrix multiplications.

	Operation Count	Standard, operation	Saving
= P	$(c^2+1)(2c+1)$	(2C+1) <sup>3</sup>	_
ŋ	$= 20^3 + 0^2 + 20 + 1$	$= 80^3 + 120^2 + 60 + 1$	$60^3 + 110^2 + 40$
j	(C+1) <sup>2</sup>	80 <sup>3</sup> + 120 <sup>2</sup> + 60+1	$80^{3} + 110^{2} + 40 + 1$
J	$= C^2 + 2C + 1$		
j	(C+1)	(20+1) <sup>2</sup>	40 <sup>2</sup> + 3 C
Ü		$= 40^2 + 40 + 1$	
$^{ ext{P}}_{ ext{n}}$	x <sub>n</sub> 20 <sup>2</sup> + 30+1	4°C <sup>2</sup> + 4°C+1	20 <sup>2</sup> +0
Inv	50 <sup>3</sup> + 140 <sup>2</sup> +60	80 <sup>3</sup> + 120 <sup>2</sup> + 60+1	30 <sup>3</sup> -20 <sup>2</sup> +1
Pj j	$(N-1)(2C^3+C^2+2C+1)$ $N(C^2+2C+1)$ $N(C+1)$	$(N-1)(8c^3+12c^2+6c+1)$ $N(8c^3+12c^2+6c+1)$ $N(4c^2+4c+1)$ $-(N-1)(4c^2+4c+1)$	
n	$x_{n} (N-1)(C^{2}+2C+1)$	(N-1)(4C + 4C+1)	
Inv	$r   N(5c^3 + 14c^2 + 6c)$	$N (8C^3 + 12C^2 + 6C + 1)$	
To	tal .3N(50 <sup>3</sup> + 150 <sup>2</sup> +90+2) unt + 2(N-1)(20 <sup>3</sup> +20 <sup>2</sup> +30-	$3N(160^3 + 280^2 + 160 + 3)$ +1) + $2(N-1)(80^3 + 160^2 + 1)$	

#### CHAPTER 3

### RESULTS AND DISCUSSION

The algorithm proposed in the previous chapter has been implemented in FORTRAN 10 on the DEC 10 system. First, the saving in CPU time for inventing the matrix  $(\bar{\bar{B}}_n - \bar{\bar{A}}_n \bar{\bar{C}}_{n-1})$  has been examined. The efficiency of the proposed method has been tested by solving two absorption problems and the details are presented in this chapter.

It is known that the inversion of matrix by partition does not result in saving the CPU time. But, the structure matrix in these separation process problems is such that we need to invert only  $C \times C$  matrix to get the inversion of  $(2C+1) \times (2C+1)$ . This is so because the other submatrix, obtained on partitioning, whose inverse to be found to found is either identity matrix or a diagonal matrix with only a few elements which are other than one. Thus some CPU time saving can be achieved. As shown in the previous Chapter, the operation count for the inversion for a problem involving C components is  $(5C^3 + 14C^2 + 6C)$  compared the standard inversion (i.e. either by the Gaussian elemination or the Gauss-Jordon elimination) which is  $(2C+1)^3$ .

TABLE 2

SAMPLE PROBLEM NO.1 Taken from [3]

Absorption column having 20 plates and 4 component systems

Component	Rich gas VN+1,i mole/hr	Absorbir liq. lo,i mol	
A	75.0	0.0	
В	15.0	0.0	Absorbing liq. temp. = 125,°C
C	10.0	0.0	Entering rich gas temp.=200°C
D ×	0.0	100.0	

Find the temperature and flow rates of the components of vapor and liquid stream leaving from the column.

TABLE 3

SAMPLE PROBLEM No.2

STATEMENT OF THE PROBLEM

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COMPONENT	RICH GAS VN+1,i	Lean Oil l <sub>o.,</sub> i	OTHER SPECIFICATIONS
MIL COMPUNING AND	(mol/h)	(mol/h)	and the second control of the second control
002.	0.4703	0.0	$T_0 = 2.9 ^{\circ}F, T_{N+1} = 0 ^{\circ}F,$
N <sub>2</sub>	0.1822	0.0	N = 8, and
CH <sub>4</sub>	88.7000	0.0	$P = 800 \text{ lb/in}^2 \text{ abs}$
<sup>C</sup> 2 <sup>H</sup> 6	6.6747	0.0	initial temperature
03 <sup>H</sup> 8	2.7786	0.0015	profile to be constant
i <sup>C</sup> 4 <sup>H</sup> 10	0.6375	0.0006	at m <sub>j</sub> = 25°F
n C <sub>4</sub> H <sub>10</sub>	0.3655	0.0013	for all $j (j = 1, 2N)$ .
i <sup>C</sup> 5 <sup>H</sup> 12	0.1158	0.0067	The initial vapor rate
n C <sub>5</sub> H <sub>12</sub>	0.0505	0.0061	profile is to be constant
<sup>C</sup> 6 <sup>H</sup> 14	0.0146	0.1495	at V = 90.88(j=1,28)
<sup>C</sup> 7 <sup>H</sup> 16	0.0081	0.5736	and the liquid rates are
<sup>C</sup> 8 <sup>H</sup> 18	0.002	1.8214	$L_j = 6.3095 (j = 1,28)$
<sup>C</sup> 9 <sup>H</sup> 20	0.000	1.6866	and $L_8 = 15.42$
<sup>C</sup> 10 <sup>H</sup> 22	0.000	2.0619	

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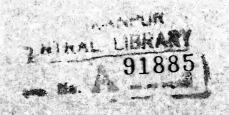
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To get actual CPU time savings, numerical experiments were carried out using the test matrix of the type that are encountered in the separation process problems. The test matrices are tabulated in Appendix (D). The CPU time required for the inversion by partition and by the Gauss-Jordon elemination have been compared for different values of C and presented in Table 1.

The saving in CPU time is becoming increasingly significant as the C increases if the proposed method is employed.

An absorption problem given by Naphthali-Sandholm 3 and another absorption problem given by Holland 1 have been chosen for testing the efficiency of the proposed method. In the first problem, the total number components of the mixture are four and the number stages are twenty. In the other problem the number components are fourteen and stages are eight. The details of the problems are given in Table 2 and 3 and the equilibrium and enthalpy data are given in Appendices B & C. The elements of the Jocabian matrix (the partial derivatives) have been evaluated analytically. The analytical expressions for the derivatives are given in Appendix (A & E).

The two problems have been solved exploiting the sparsity of the submatrices and employing the 'standard' matrix operations. Since no approximations were made while taking of the sparsity in computation, the number of iteration required identical in both the methods. The CPU time required to solve the two problems by these two methods alongwith other details are given in Table 4.

Table 4 shows a substantial reduction in CPU time is achieved in the proposed method. The saving in CPU time becomes increasingly significant as the number of components increases. The proposed method can be employed even interlinked columns.

#### CHAPTER 4

#### CONCLUSIONS

An efficient algorithm for solving the separation processes problems by the well known Naphthali-Sandholm method has been presented. In this algorithm the sparsity the submatrices of the Jacobian matrix is exploited in the matrix multiplications and in the inversion of the matrices  $(\tilde{B}_n - \tilde{A}_n \tilde{C}_{n-1})$ . It has been shown that the operation count for the proposed algorithm is  $(50^3 + 140^2 + 60)$  compared to  $(20+1)^3$  with the standard matrix operations. By solving two 'bench-mark' problems, it has been shown that the saving in the CPU time becomes increasingly significant as the number components involved becomes large. In the proposed algorithm, the saving in CPU time in computing the correction vector  $\tilde{A}_n$  is effected, but the number attrix operations.

The algorithm can be extended for solving interlinked column employing the method suggested by Hocling and Seader.

The computer code need to be tested to determine its effectiveness for distillation of nonideal mixtures, extraction and adsorption problems.

## NOMENCIATURE

= A _j	Submatrices of Jacobian Matrix at the jth row
=° B =j	Submatrices of Jacobian Matrix at the jth row
= ° ;	Submatrices of Jacobian Matrix at the jth row
Q	Total number of components involved
Lj	Overall molar liquid flow rate from jth stage
v <b>j</b> i	Overall molar Vapor flow rate from jth stage
l	Molar liquid flow rate of component i from jth stage
v ij	Molar vaper flow rate of component i from jth stage
K ij	Distribution coefficient of ith component at jth stage
$\overline{\mathbf{X}}$	Vector of variables
J	Jacobian matrix
f	Molar feed rate of the component i into the jth stage
F	Residual vector
F =j	Residual vector for the 3th stage
=° P j	P matrix in the Thomas algorithm
q j	External heat input into the jth stage
h	Molar specific liquid phase enthalpy of component i
H <sub>i</sub>	Molar specific vapor phase enthalpy of component i
h <sub>fi,j</sub>	Molar specific feed enthalpies of component i
N	Total number of stages
<sup>M</sup> ij	Mass balance discrepancy of component i for jth stage

O<sub>ij</sub> Equilibrium relation discrepancy function for component i at jth stage

E<sub>j</sub> Enthalpy balance discrepancy function for the jth stage

S<sub>j</sub> Fraction of the vapor stream withdrawn from the jth stage

s<sub>j</sub> Fraction of the liquid stream withdrawn from the jth stage

## Greek letters

difference in variable

murphree efficiency of jth stage

Kronecker delta function

partial derivative

activity coefficient of component i

Thermodynamic parameter

P matrix of the Thomas Algorithm

q matrix in the Thomas Algorithm

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TA BLE 1

CPU time for the inversion of matrix by the proposed method and the Gauss Jordan Elimination

Ratio of	tunoo		92.0	0.70	0.68	99*0	
Ratio of	(IP/GJ)		0.56	0.43	0.35	15°0	
time (in Secs.)	Gauss Jordan Elimination (GJ)		0.25	1.38	5.54	16.22	
CPU time (i	Inversion by partition(I.P)	ELEMENT I STORAGE AND THE COMMENT OF	0.14	0.59	7		4• <i>yy</i>
No. of components			72	70	) - T	o 6	0.2

TABLE 4

CPU time for two different problems

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GPU in Normal t <sub>1</sub>	7.93	784.64 784.62 784.62	
Rum No.	-25	70N	
No. of iterations needed I	N	ſΛ	
No. of stages	20	ω	,
No. of components	4	14	
Problem No.	1	01	

#### APPENDIX A

### ELEMENTS OF THE SUBMATRICES

### 1. FOR IDEAL VIE SYSTEM

# A. FOR A MATRIX

For 
$$1 \le i$$
,  $j \le C & 1 \le n \le N$ 

$$\frac{\partial M_{i,n}}{\partial l_{j,n-1}} = S_{i,j}, \quad \frac{\partial M_{i,n}}{\partial v_{j,n-1}} = \frac{\partial M_{i,n}}{\partial T_{n-1}} = 0$$

$$\frac{\partial o_{i,n}}{\partial l_{j,n-1}} = \frac{\partial o_{i,n}}{\partial v_{j,n-1}} = \frac{\partial o_{i,n}}{\partial T_{n-1}} = 0$$

$$\frac{\partial E_n}{\partial l_{i,n-1}} = h_{i,n-1}; \frac{\partial E_n}{\partial v_{i,n-1}} = O$$

$$\frac{\partial E_{n}}{\partial T_{nr}} = \sum_{\substack{i=1 \ i,n-1}}^{C} 1_{i,n-1} \frac{\partial h_{i,n-1}}{\partial T_{n-1}}$$

# B. FOR B MATRIX

For  $1 \le i$ ,  $j \le C$  &  $1 \le n \le N$ 

$$\frac{\partial M_{i,n}}{\partial l_{j,n}} = - (1 + s_n) \delta_{ij}$$

$$\frac{\partial M_{i,n}}{\partial v_{i,n}} = - (1 + S_n) \stackrel{\leq}{\downarrow}_{ij}$$

$$\frac{\partial M_{i,n}}{\partial T_n} = 0$$

Appendix A continued.

$$\frac{\partial O_{i,n}}{\partial I_{j,n}} = \eta_n k_{i,n} \qquad \frac{\delta_{ij} I_n - I_{in}}{L_n^2}$$

$$\frac{\partial O_{i,n}}{\partial V_{j,n}} = \frac{v_{i,n} - v_n \delta_{ij}}{v_n^2}$$

$$\frac{\partial O_{i,n}}{\partial T_n} = \frac{\eta_n I_{i,n}}{V_n^2} \qquad \frac{dk_{i,n}}{dT_n}$$

$$\frac{1 \cdot n}{\sqrt{T_n}} = \frac{n \cdot 1 \cdot n}{I_n} \frac{1 \cdot n}{dT_n}$$

$$\frac{E_n}{\sqrt{I_{i,n}}} = - (1 + s_n) h_{i,n}$$

$$\frac{\partial E_{n}}{\partial v_{i,n}} = -(1+S_{n}) H_{i,n}$$

$$\frac{\partial E_{n}}{\partial T_{n}} = -\sum_{i=1}^{n} (1+S_{n}) v_{i,n} \frac{dH_{i,n}}{dT_{m}} - \sum_{i=1}^{n} (1+S_{n}) l_{i,n} \frac{dh_{i,n}}{dT_{n}}$$

C. FOR O MATRIX

$$\frac{\partial M_{i,n}}{\partial T_{j,n+1}} = 0; \quad \frac{\partial M_{i,n}}{\partial V_{j,n+1}} = \int_{ij}; \quad \frac{\partial M_{i,n}}{\partial T_{n+1}} = 0$$

$$\frac{\partial O_{i,n}}{\partial T_{i,n+1}} = 0; \quad \frac{\partial O_{i,n}}{\partial V_{i,n+1}} = (1 - \gamma_m) \frac{\int_{ij} V_{n+1} - V_{i,n+1}}{V^2}$$

Appendix A continued

$$\frac{\partial O_{i,n}}{\partial T_{n+1}} = O = \frac{\partial E_n}{\partial T_{i,n+1}}$$

$$\frac{\partial E_{n}}{\partial v_{i,n+1}} = H_{i,n+1} ; \quad \frac{\partial E_{n}}{\partial T_{n+1}} = \sum_{i=1}^{C} v_{i,n+1} \frac{dH_{i,n+1}}{dT_{n+1}}$$

# APPENDIX B

# SAMPLE PROBLEM No.1

AT AC

K values

Material	Temperatu	T C
Albiria (Amministrus), Maris (Amris Treasanisms) (1894, 11 America)	100°F .	200°F
A	500.0	550.0
B ,	1.50	1.8
C	0.90	1.00
D	$1.0 \times 10^{-6}$	1.5 x 10
omprimienta ja 1966. kähilyyksikkii ja 2004. juutaksikkiilija ja kaleisikkiilija ja kaleisikkiilija.  Markinsi liittiin – hakkiili kaleisikkiilija, ja markin – 1865 – 1860. ili 2004 –	Molar liquid enthalpi	es, 10 <sup>3</sup> cal/mol
A	0.01	0.013
В	0.30	0.33
C	0.40	0.44
D	1.50	1.90
Aurel, M. (-paper 20) — Affect Affect 27 (2000-) confined followid. Telescolored	Molar vapor enthalpie	s, 10 <sup>3</sup> cal/mole
A.	1.00	1.002
В	1.80	1.82
Q	2.00	2.03
D	5.75	5.95

- 25°F to 40°F and at a pressure of 800 lb/in abs. K values for the temperature range of

componen t	ali	a2i	431	41
00,	- 0.62822223x10 <sup>-1</sup>	0.30688802x10-3	0.39996468x10 <sup>-6</sup>	-0.57899830x10-9
N CN	0.50596821	-0.43488364x10 <sup>-3</sup>	-0.15009991x10 <sup>-5</sup>	0.34494154×10 <sup>-8</sup>
CH,	0.15584934	-0.15205775x10 <sup>-3</sup>	0.50349212x10 <sup>-6</sup>	-0.17713546x10 <sup>-9</sup>
CoH <sub>C</sub>	0.91486037x10 <sup>-1</sup>	-0.16355944x10-3	0.33741924x10 <sup>-6</sup>	0.14797150x10 <sup>-9</sup>
$^{\circ}_{z^{\mathrm{H}_{\mathrm{S}}}}$		-0.64491702x10-4	0.29233627x10 <sup>-6</sup>	-0.48597680x10-11
i C,H,O		-0.94310963x10 <sup>-4</sup>	0.28026648x10 <sup>-6</sup>	0.10462797x10-10
n C <sub>4</sub> H <sub>4</sub> O		-0.13635085x10 <sup>-3</sup>	0.37584653x10 <sup>-6</sup>	-0.69237741x10 <sup>-10</sup>
T CEH12		-0.34736106x10 <sup>-4</sup>	0.12591028x10 <sup>-6</sup>	0.73157133x10 <sup>-10</sup>
1 CH10		-0.40284984x10-4	0.14439195x10 <sup>-6</sup>	0.56656790x10-10
n C <sub>E</sub> H <sub>1</sub> 1	0.88765752x10 <sup>-3</sup>	0.37082646x10 <sup>-4</sup>	-0.40746951x10-7	0.15187203x10 <sup>-9</sup>
n 0,H,6	0.63677356x10 <sup>-2</sup>	-0.64409760x10 <sup>-5</sup>	0.31793974x10 <sup>-7</sup>	0.78284379x10-10
$n c_{\mathrm{gH}_{18}}$		-0.34673591x10-4	0.82305291x10-7	0.21022392x10 <sup>-10</sup>
n C <sub>Q</sub> H2O	0.78793392x10-2	-0.2	0.62435951x10 <sup>-7</sup>	0.25793478x10-10
n C <sub>10</sub> H <sub>22</sub>	0.64146556x10 <sup>-2</sup>	0.1	0.30005250x10 <sup>-7</sup>	0.30266026x10-10

 $K_{i} = T (a_{1i} + a_{2i} T + a_{3i} T^{2} + a_{4i} T^{3})^{3} (T in ^{R})$ 

0.84406456x10<sup>-5</sup> 0.42183183x10<sup>-6</sup> 0.18765335x10-4 0.14662071x10-4 0.12580301x10-4 0.89584664x10<sup>-5</sup> 0.16110935x10-4 0.25869682x10<sup>-4</sup> 0.19637475×10-4 0.60245657x10<sup>-4</sup> 0.42837073x10<sup>-5</sup> 0.70521889x10<sup>-6</sup> 0.83943509x10-4 0.75087155x10<sup>-4</sup> ъ 41 0.79399594x10<sup>-2</sup> 0.15354034x10-1 0.31726830x10-1 0.38262386x10-1 0.43731511x10-1 0.38164884x10<sup>-1</sup> 0.40186413x10-1 -0.79501927x10<sup>-1</sup> -0.58315706x10<sup>-1</sup> 0.30206113x10<sup>-1</sup> 0.54925641x10<sup>-1</sup> -0.67470074x10<sup>-1</sup> 0.17655857x10-1 0.2791080x10<sup>-1</sup>  $^{b}$ 3i 0.18159073x10<sup>2</sup> 0.19229950x10<sup>2</sup> 0.81615143×10<sup>2</sup> 0.74108551x10<sup>2</sup> 0.70354599×10<sup>2</sup> 0.11454294×10<sup>2</sup> 0.10596339x10<sup>2</sup> 0.46583524x10<sup>1</sup> 0.92456856x10<sup>1</sup> 0.66175545x10<sup>1</sup> 0.39552670x10<sup>1</sup> 0.7206460x10<sup>1</sup> 0.3731512x10<sup>1</sup> 0.5446243x10  $^{\mathrm{b}_{2i}}$ -0.10632578x10<sup>2</sup> -0,67810352x10<sup>4</sup> -0.44150469x10<sup>4</sup> -0.79141992x10<sup>4</sup> 0.97404712x10<sup>3</sup> 0,33611663x10<sup>4</sup> 0.66707016x10<sup>2</sup> 0.81635181x10<sup>3</sup> 0.43454375x104 0.15837112x10<sup>4</sup> 0.21237510x10<sup>4</sup> 0.17543628x10<sup>4</sup> 0,32309192x10<sup>4</sup> 0.22524075x10<sup>4</sup> Component 1 C4H10  $^{\mathrm{n}}$   $^{\mathrm{G}_{10}}$  $^{\mathrm{n}}\,^{\mathrm{c}}_{9}{}^{\mathrm{H}}_{20}$ n C<sub>8</sub>H<sub>18</sub>  $^{N_2}_{\text{CH}_4}$ 

800 lb/infabs

[[

a t

- 25°₽ to 40°F

THE TEMPERATURE RANGE OF

LIQUID ENTHALPIES FOR

(Tin oR) Btu/lb mole  $h_i = b_{1i} + b_{2i} I + b_{3i} I^2 + b_{4i} I^3$ 

To  $40^{\circ}$ F AT P = 800 lb/in<sup>2</sup>abs. A PPENULX C CONTINUED: VA POR ENTHALPIES FOR THE TEMPERATURE RANGE OF - 25°F

	, t	~2i	31	41
00°	0.13978977×10 <sup>5</sup>	- 0.96359463x10 <sup>1</sup>	0.38228422x10 <sup>-1</sup>	-0.26870170x10 <sup>-4</sup>
$N_{2}$	0.48638672x10 <sup>4</sup>	- 0.21227379x10 <sup>1</sup>	0.17565668x10 <sup>-1</sup>	-0.11367006x10 <sup>-4</sup>
CH	0.63255430x104	$-0.20747757x10^{1}$	0.18532634x10 <sup>-1</sup>	-0.10630416x10 <sup>-4</sup>
3HcD	0.10628934x10 <sup>5</sup>	- 0.28718834x10 <sup>1</sup>	0.24877094x10 <sup>-1</sup>	-0.13233222x10 <sup>-4</sup>
$G_{\chi}H_{\Sigma}$	0.13954383x10 <sup>5</sup>	$-0.41930256x10^{1}$	0.32614145x10-1	-0.15483340x10 <sup>-4</sup>
i CAH,	0.94088984x104	0.39262680x10 <sup>2</sup>	-0.55596594x10 <sup>-1</sup>	●.51507392x10 <sup>-4</sup>
n CAHAO	0.57302344x104	0.75117737x10 <sup>2</sup>	-0.13120884x10 <sup>0</sup>	0.10517908x10 <sup>-3</sup>
1 C <sub>E</sub> H <sub>1</sub> 2	0.83081953x10 <sup>4</sup>	0.75267792x10 <sup>2</sup>	_0.12945843x10 <sup>0</sup>	0.10845697x10 <sup>-3</sup>
n C <sub>E</sub> H <sub>12</sub>	0.12804211x10 <sup>5</sup>	0.61654007x10 <sup>2</sup>	-0.97365201x10 <sup>-1</sup>	0.84398722x10 <sup>-4</sup>
n C <sub>H</sub> <sub>1</sub>	0.23001684x10 <sup>5</sup>	$0.27744919x10^{2}$	-0.31545494x10 <sup>-1</sup>	0,49981289x10 <sup>-4</sup>
n C <sub>7</sub> H <sub>7</sub> 6	0.14876816x10 <sup>5</sup>	0,59342438x10 <sup>2</sup>	-0.81853271x10 <sup>-1</sup>	0.81429855x10 <sup>-4</sup>
n C <sub>B</sub> H <sub>18</sub>	0,32793215x10 <sup>5</sup>	- 0.35040283x10 <sup>2</sup>	0.11162955x10 <sup>0</sup>	-0.42647429x10 <sup>-4</sup>
n C <sub>9</sub> H <sub>2O</sub>	0.47024656x10 <sup>5</sup>	- 0.95395035x10 <sup>2</sup>	0.24547529x10 <sup>0</sup>	-0.13209638x10 <sup>-3</sup>
n C <sub>10</sub> H <sub>22</sub>	0.55238211x10 <sup>5</sup>	- 0.13195618x10 <sup>3</sup>	0.32518369x10 <sup>0</sup>	-0.18188384x10 <sup>-3</sup>

( T in 'R) Btu/lb mole.  $H_{i} = C_{1i} + C_{2i} T + C_{3i} T^{2} + C_{4i} T^{3}$ 

00000  Appendix D (continued)

### APPENDIX E

# ELEMENTS OF MATRICES FOR NONIDEAL SYSTEM

For nonideal system only the following elements need to evaluate from different expressions. Following expressions are evaluated using UNIQUAK model. Because of the lack of time and for model independent programme, this feature was not incorporated.

$$\frac{\partial o_{i,n}}{\partial v_{j,n}} = \frac{\delta_{ij} v_{n} - v_{i,n}}{v_{n}^{2}} + \frac{1}{L_{n}} \frac{\sqrt{L_{i,n}}}{\sqrt{V_{i,n}}} 2 \frac{\sqrt{L_{i,n}}}{\sqrt{V_{i,n}}}$$

$$\frac{\partial o_{i,n}}{\partial l_{i,n}} = -\frac{1}{\sqrt{V_{i,n}}} \frac{1_{i,n}}{L_{n}} \frac{\sqrt{V_{i,n}}}{\sqrt{V_{i,n}}} + \sqrt{L_{i,n}} \frac{\sqrt{L_{i,n}}}{L_{n}^{2}}$$

$$\frac{\partial o_{i,n}}{\partial l_{i,n}} = -\frac{1_{i,n}}{L_{n}} \frac{\sqrt{V_{i,n}}}{\sqrt{V_{i,n}}} \frac{\sqrt{V_{i,n}}}{\sqrt{V_{i,n}}} + \ln\left(\frac{v_{i,n}}{\sqrt{V_{i,n}}}\right) \frac{\sqrt{V_{i,n}}}{\sqrt{V_{i,n}}}$$

$$\sqrt{V_{i,n}} = \exp\left[1 - \frac{v_{i,n}}{\sqrt{V_{i,n}}} + \ln\left(\frac{v_{i,n}}{\sqrt{V_{i,n}}}\right) - \frac{v_{i,n}}{\sqrt{V_{i,n}}} + \ln\left(\frac{v_{i,n}}{\sqrt{V_{i,n}}}\right) + \frac{v_{i,n}}{\sqrt{V_{i,n}}} + \ln\left(\frac{v_{i,n}}{\sqrt{V_{i,n}}}\right) + \frac{v_{i,n}}{\sqrt{V_{i,n}}} + \frac{v_{i,n}}{\sqrt{V_{i,$$

Appendix E continued.

Similarly for  $\gamma_{i,n}^{L}$  where v is substituted by 1

$$\frac{2\gamma_{i,n}^{\prime}}{2^{\prime}v_{j,n}} = \frac{1}{\sqrt{i,n}} \left[ -\frac{r_{i}}{\sqrt{k}} \frac{v_{k,n} r_{k} - v_{n} r_{j}}{\sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k}} + \frac{v_{k,n} r_{k} - r_{j} v_{m}}{\sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k}} + \frac{v_{k,n} r_{k} - r_{j} v_{m}}{\sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k} \sqrt{k}} - \frac{z}{2} q_{i} \right] - \frac{r_{i}}{q_{i}} \frac{(\sum_{k} v_{k,n} r_{k}) q_{j} (\sum_{k} v_{k,n} r_{k})}{(\sum_{k} v_{k,n} r_{k})} + \frac{q_{j} \sum_{k} (v_{k,n} r_{k}) - r_{j} \sum_{k} w_{k,n} q_{k}}{(\sum_{k} v_{k,n} r_{k})} - \frac{q_{j} \sum_{k} (v_{k,n} q_{k}) \sum_{m} m_{j} q_{m} A_{j,m} - q_{j} \sum_{m} q_{m} A_{m,i} v_{m,n}}{(\sum_{m} (v_{m,n} q_{m} A_{j,m}) \sum_{k} (v_{k,n} q_{k})} - \frac{q_{k} \sum_{m} (v_{m,n} q_{m} A_{k,m}) - v_{k,n} q_{j} A_{k,j}}{(\sum_{m} (v_{m,n} q_{m} A_{k,m}) - v_{k,n} q_{j} A_{k,j}} + \frac{v_{i,n} q_{m} A_{k,m}}{v_{n}} - \frac{1}{v_{i,n}} \frac{v_{i,n}}{v_{i,n}} + \frac{v_{i,n} q_{m} A_{k,m}}{v_{i,n}} + \frac{v_{i$$

Appendix E continued.

$$\frac{\partial \gamma_{i,m}}{\partial T_{n}} = \sqrt{\frac{1}{i,n}} \left\{ -\frac{q_{ij}}{\partial T_{n}} \right\} \frac{\partial \Lambda_{i,i}}{\partial T_{n}} - \frac{\int_{K} e_{j} \Lambda_{k,j} e_{k} \partial T_{n}}{\int_{K} e_{j} \Lambda_{k,j} e_{k} \partial T_{n}} \right\}$$

$$\frac{-e_{k} \Lambda_{k,j}}{\int_{K} e_{j} \Lambda_{i,j}} \frac{\partial \Lambda_{k,j}}{\partial T_{m}}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{i,j} e_{k} \partial T_{n}}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{n,j}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} - e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{n,j}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{i,j}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{i,j}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{i,j}} e_{k} \frac{\partial \Lambda_{k,j}}{\partial T_{n}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} \Lambda_{i,j}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{k}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}} e_{i}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_{i,j}}{\int_{K} e_{j} A_{i,j}}$$

$$= -\frac{1}{RT_{m}^{2}} \frac{e_{i} \sum_{j} e_{j} \Lambda_$$

```
GENERAL PROGRAM FOR DESIGN AND SIMULATION OF MULTICOMPONENT MULTISTAGE EQUILIBRIUM SEPARATION PROCESSES

DEVELOPED AND PROGRAMMED BY L. FIENK

EXTENDED NAPHTHALI SANDHOLM METHOD SPARSITY AND SYMMETRY EXPLOITATION.
LISTING OF SYMBOLS ARE AS FOLLOWS

AL(J):LIO. RATE OF THE J:TH STAGE:

V(J):VAP RATE OF THE J:TH STAGE:

SV(J,I): VAP. RATE OF COMPONENT I AT THE J TH STAGE

SV(J,I): VAP. RATE OF COMPONENT I AT THE J TH STAGE

SL(J):HEAT INPUT AT THE J TH STAGE:

SS(J):FRACTION OF VAP. STREAM TAXEN FROM THE J TH STAGE

SSS(J):FRACTION OF VAP. STREAM TAXEN FROM THE J TH STAGE:

SSS(J):FRACTION OF VAP. STREAM TAXEN FROM THE J TH STAGE:

SH(I): LIO PHASE ENTHALPY OF COMPONENT I AT ANY STAGE

SH(I): LIO PHASE ENTHALPY OF COMPONENT I AT ANY STAGE:

H(I): LIO PHASE ENTHALPY OF COMPONENT I AT J. TH STAGE:

H(I): ENTHALPY OF FEED AT ANY STAGE FOR COMPONENT I

H(I): ENTHALPY OF COMPONENT I NEXT BOITOM STAGE OF THE STAGE

TO: TEMPERATURE OF THE LIQUID ENTERING INTO TIRST PLATE:

SLO(I): LIO. RATE OF COMPONENT I ENTERING INTO TIRST PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL VAPOR RATE ENERING INTO THE NT PLATE:

VB : TOTAL NUMBER OF COMPONENT I NEXT UP. STAGE OF THE STAGE.

C: TOTAL NUMBER OF COMPONENT I AT THE J TH STAGE

C: TOTAL NUMBER OF COMPONENT I AT THE J TH STAGE

O: ENTHALPY BALNCE NORMALISATION FACTOR
   LISTING OF SYMBOLS ARE AS FOLLOWS
   INTEGER C,CT
COMMON C,CT,N
COMMON /AREA1/SLO,SVB,TB,TO,VB/AREA2/SS,SSS,TF,ETA,SQ
COMMON /A1/SA/A2/SC/A3/SE
DIMENSION B(29,29),C1(29,29),PB(14),PC(14),C2(8,29,29)
```

```
*)TB,TO,VB

*)(($A(1,J),J=1,4),I=1,C)

*)(($C(I,J),J=1,4),I=1,C)

*)(($E(I,J),J=1,4),I=1,C)
                   EAD(41,
EAD(41,
EAD(41,
READ(41,*)((SC(I,J),J=1,4),I=1,Z)

READ(41,*)((SE(I,J),J=1,4),I=1,Z)

END OF INPUT SPECIFICATION

AEF=0.

IF(IN.GE.1)GOTO 701

WRITE(44,2210)

FORMAT(4X,Z11)

FORMAT(4X,Z11)

FORMAT(4X,Z12)

FORMAT(4X,Z12)

FORMAT(4X,Z13)

FORMAT(4X,Z13)
```

```
VRITE(44,332)(I,(SL(J,I),J=1,N),I=4,C)

VRITE(44,4440)

FORMAT(//,'INITIAL VAPOR AND LIQUID RATE AND TEMPERATURE GUESS

VRITE(44,4467)

VRITE(44,4468)

VRITE(44,4610)

VRITE(44,6100)

VRITE(44,6100)

VRITE(44,6100)

VRITE(44,6100)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6101)

VRITE(44,6102)

VRITE(44,6102)

VRITE(44,6102)

VRITE(54,6102)

VRITE(5,15)

FORMAT(27X,'1984*)

VRITE(5,15)

FORMAT(27X,'1984*)

VRITE(5,15)

FORMAT(4X,'DEVELOPED AND PROGRAMMED BY L. FIENK*)

VRITE(5,15)

FORMAT(/,4X,'DEVELOPED AND PROGRAMMED BY L. FIENK*)
           TE(5,13)
MAT(7,4x, DEVELOPED AND PROGRAMMED BY L. FIENK")
COMPUTATION OF AUGMENTED ERROR FUNCTION AT THE GUESS VECTOR
# # CONTRIBUTION DUE: TO MASS BALANCE # # #
START OF EVALUATION OF THE C
ENTHALPY BALNACE:
```

```
D02 J=1,N-1
TT=T(J)
P1=1+SS(J)
P1=1+SS(J)
P1=1+SS(J)
P1=1+SS(J)
P1=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1+SS(J)
P2=1-SS(J)
P2=1-SS(J)
P2=1-SS(J)
P2=1-SS(J)
P2=1-SS(J)
P3=1-SS(J)
P3
                               TO COMPUTE THE LIQUID & VAPOR FLOW RATE FOR ALL STAGES

TO COMPUTE THE LIQUID & VAPOR FLOW RATE FOR ALL STAGES

TO CONTRIBUTION DUE TO EQUILIBRIUM CONDITION CONSIDERING

EFFICIENCY OF ALL STAGES
```

```
B(2*C+1,2*C+1)=B(2*C+1,2*C+1)*P1+FT*P2:
B(CT,Cf)=B(CT,CT)*Q
D(1,1)=-P2
B(I,T)=-P1
CONTINUE
B(I,T)=-P1
CONTINUE
B(I,T)=-P1
CONTINUE
B(I,T)--P1
CONTINUE
B(I,T)

                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                       ERROR VECTOR*,/(4X,9(E13.6,3X)))
               DO 54 J=2,N
P1=1.+SS(J)
P2=1.+SSS(J)
TT=T(J=1)
CALL DENL(TT,H)
AA=0.
CALL ENL(TT,SHH)
DO 55 I=1,C
AA=AA+SL(J=1,I)*H(I)
A(CT,I)=SHH(I)*Q
A(CT,I)=SHH(I)*Q
CONTINUE
```

```
A(CT,CT)=AA*O
END OF COMPUTATION OF ELEMENTS OF A MATRIX
WRITE(5,307)(SA(I4)); J=1
PORMAT(Y,Y,4X,EL4ENTS)
PORMAT(Y,4X,EL4ENTS)
PO
```

```
DO 160 II=1,CT

A(I,II)=0.0.
CINTINUE (I) GOTO 70

FT=T(J+1)=0.N) GOTO 70

CALL ENVITT,PA)

DO 56 I=15C
CONTINUE (I)=1.N)
CALL DENV(TT,PA)

DO 59 I=1.SC (I)=1.CC
CONTINUE 301 (I)=1.CT)

DO 87 II=1.CC
CI(I+C,II)=0.N) GOTO 88

CI(I+C,II)=0.N) GOTO 9.0

CI(I+C,II
```

```
OF=-1.

DO 135 J=1.N

DO 136 I=1.C

SL(J,I)=SL(J,I)-DF*F1(J,I)C

SV(J,I)=SV(J,I)-DF*F1(J,I+C)C

CONTINUE

T(J)=T(J)-OF*F1(J,CT)

CONTINUE

IN=IN+1

WRITE(44,200)IN

FORMAT(/,"ITERATION NUMBER=",I3)

AEF=0.

DO 25 J=1.N

V(J)=0.

CDNTINUE

DO 202 I=1.CT

DO 202 I=1.CT

B(I,II)=0.

CONTINUE

CONTINUE
     DO 202 II=1.CT

C1(I,II)=0.

C1(I,II)=0.

C1(I,II)=0.

C1(I,II)=0.

C371 J=1,N

C1(J)+$V(J,I)

AL(J)=AL(J)+$V(J,I)

CONTINUE

CDNTINUE

CD TO 16

NRITE(44,9)

NRITE(44,20)IN

FORMAT(4,'ITERATION NUMBER=',I3)

WRITE(44,2410)

FORMAT(4x,*FINAL VALUES OF FLOW RATES AND TEMPE WRITE(44,2411)

FORMAT(4x,*VAPOR PHASE CDMPONENT FLOW RATES AF WRITE(44,2413)

FORMAT(4x,36('-'))

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2414)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

FORMAT(4x,13('-'),10x,14('-'),/)

WRITE(44,2413)

WRITE(44,2413)

WRITE(44,2413)

WRITE(44,392)(I,(SV(J,I),J=1,N),I=1,C)

WRITE(44,2413)

WRITE(44,2413)
                                                                                                                                                                                                                                                                         FLOW RATES AND TEMPERATURE ARE: ")
                                                                                                                                YAPOR PHASE COMPONENT FLOW RATES ARE')
                                                                                                                                        ,12X,'2",12X,'3",12X,'4',12X,'5',12X,'6",14X,
```

```
FORMAT(//,4X, 'VAPOR AND LIQUID RATE AND TEMPERATURE AND REMETE (44,4467)
FORMAT(4X,42('-'))
WRITE(44,4468)
FORMAT(4X, 'STAGE NUMBER',4X, 'VAPOR RATE ',4X,

1 RATE TEMPERATURE ')
WRITE(44,7979)
FORMAT(4X,12('-'),4X,19('-'),4X,17('-'),4X,19('-'),/)
WRITE(44,13)(J,V(J),AL(J),T(J),J=1,N)
FORMAT(8X,13,5X,E19.8,4X,E17.8,4X,E19.8)
IFAIL=1
STOP
END
SUBROUTINE MTMUL2(A,B)
INTEGER C,CT
COMMON C,CT,N
DIMENSION A(CT,CT),B(CT,CT),D(29,29)
DO 1 I=1,CT
DO 1 J=1,CT
DO 2 J=1,CT
DO 2 J=1,CT
B(I,J)=D(I,J)
                                                       VAPOR AND LIQUID RATE AND TEMPERATURE ARE (1/1)
                                                                                                                                                                                                   *,4X,*
                                                                                                                                                                                                                                     LIQUID
```

```
RETURN

SUBROUTIME ENV(F,EV)

UTEGER C.CTN

COMMON /A3/5E

DIMENSION EVCC), SE(14,4)

DO 1 1=1

EV(1)=SE(1,1)+SE(1,2)*T+SE(1,3)*T^2+SE(1,4)*T^3

CONTINUE

RETURN

END

SUBROUTINE DENV(T,EV)

INTEGER C.CTN

COMMON /A3/5E

DO 1 E = 1

COMMON (A3/5E)

DO 1 E = 1

COMMON (A3/5E)

DO 1 E = 1

COMMON (A3/5E)

DO 1 E = 1

CONTINUE

RETURN

SUBROUTINE SUBMAT(B,CA)

INTEGER C.CT

COMMON C.CT, CT (CT,CT)

DO 1 I = 1

CT

COMMON C.CT, CT (CT,CT)

DO 1 I = 1

CT

SUBROUTINE DIST(T,AK)

INTEGER C.CT

COMMON /A1/5A

SUBROUTINE DIST(T,AK)

INTEGER C.CT

COMMON /A1/5A

COMMON /A1/5A

CONTINUE

RETURN

END

SUBROUTINE DDIST(T,AK)

INTEGER C.CT

COMMON /A1/5A

COMMON /A1/5A

DO 1 L=1

COMMON /A1/5A

COMMON /A1/5A

DO 1 L=1

COMMON /A1/5A

COMMON /A1/5A

DIMENSION AK(C),SA(14,4)

DO 1 L=1

COMMON /A1/5A

COMMON /A1/5A

DIMENSION AK(C),SA(14,4)

DO 1 L=1

COMMON /A1/5A

COMMON /A1/5A

COMMON /A1/5A

DIMENSION AK(C),SA(14,4)

DO 1 L=1

COMMON /A1/5A

COMMON /A1/5A

COMMON /A1/5A

COMMON /A1/5A

DIMENSION AK(C),SA(14,4)

DO 1 L=1

COMMON /A1/5A

C
```

```
CONTINUE
DD 2 I=1, C
DD 2 J=1, C
DD 3 J=1, C
CONTINUE
DD 3 J=1, C
DD 4 I=1, C
DD 5 J=1, C
DD 6 I=1, C
DD 7 J=1, C
DD 7 J=1, C
DD 7 J=1, C
DD 8 S1 (I, J) + B(I+C, J+C) - BS1(I, J)
COLL, THE (BS1)
DC 6 J=1, C
DC 7 J=1, C
DC
```

```
,C),TT(14,14),UI(14,14),T(14,14),ETA(14,14)
```

```
DO 5 L=1, C
PA(1)=0.
PA(1)=0.
PA(1)=0.
B(1)=0.
D1 10 L=1(C
D1 10 L=1(C
D1 10 L=1-1(C, II)*B(II, CI)*B(II, II+C)*B(II+C, CT)
PA(II=PA(II)*B(II, II)*B(II, CI)*B(II+C, III+C)*B(II+C, CT)
D1 10 L=1 (C, II)*PA(II)*B(II, II+C)*PB(II)
D2 (10 L=1 (C, II)*B(II, II)*B(II, II+C)*B(III+C, III+C)*B(III+C, III)*B(III, III+C)*B(III+C, III+C)*B
```

```
SUBROUTIC, CT AFMUL(A,F)

INTEGEN CO,CT (CT,CT),F(CT)

DO TILL, CT,CT),F(CT)

DO TILL, CT,CT,CT),F(CT)

END TILL, CT,CT,CT)

AAANUAA +F(CT)*A(CT,CT)

AAANUAA +F(CT)*A(CT,CT)

AAEIO ROEN CO,CT (CT,CT)

AAEIO ROEN CO,CT (CT,CT)

CO,CT (CT,CT)

AAINUA +F(CT),CT (CT,CT)

AAINUA +F(CT,CT),CT (CT,CT)

CO,CT (CT,CT)

AAINUA +F(CT,CT),CT (K,J+CO)

CO,CT (CT,CT)

CO,CT (CT
```

```
150-2 35-01 27E-03 96
170-1 5E-01 62E-03 93
160-17 77E-01 1E-02 91
150-17 78E-01 1E-02 91
150-17 98E-01 3E-02 89
150-17 1 4E-02 89
150-17 1 7E-02 88
150-17 1 7E-02 88
150-17 1 7E-02 88
150-17 1 15E-01 88
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124. 197. 2
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0. 0. 0. 0
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200.
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GENERAL PROGRAM FOR DESIGN AND SIMULATION OF MULTICOMPONENT MULTISTAGE EQUILIBRIUM SEPARATION PROCESSES
               PROGRAMMED BY L. FIENK
   NAPHTHALI SANDHOLM METHOD

LISTING OF SYMBOLS ARE AS FOLLOWS

AL(J):LIO. RATE OF THE J TH STAGE!

V(J):VAP RATE OF THE JT H STAGE!

SV(J,I): VAP. RATE OF COMPONENT I AT THE J TH STAGE!

SV(J,I): LIO. RATE OF COMPONENT I AT THE J TH STAGE!

SS(J):FRACTION OF VAP. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SSS(J):FRACTION OF LIO. STREAM TAKEN FROM THE J TH STAGE!

SH(I): LIO. PHASE ENTHALPY OF COMPONENT! I AT ANY STAGE!

SH(I): LIO. PHASE ENTHALPY OF COMPONENT! I AT ANY STAGE!

HH(I): ENTHALPY OF COMPONENT I NEXT BOTTOM STAGE OF THE STAGE!

TO : TEMPERATURE OF THE LIQUID ENTERING. INTO FIRST PLATE!

SLO(I): LIO. RATE OF COMPONENT! I ENTERING INTO N. IH PLATE!

VB : TOTAL VAPOR RATE ENERING. INTO THE N. TH PLATE!

VB : TOTAL VAPOR RATE ENERING. INTO THE N. TH PLATE!

UNDER CONSIDERATION

SH(I): ENTHALPY OF COMPONENT I NEXT UP. STAGE OF THE STAGE!

UNDER CONSIDERATION

AK(I): EOILIBRIUM CONSTANT VALUE! FOR COMPONENT I AT ANY STAGE

C: TOTAL NUMBER OF COMPONENTS INVOLVED FOR SEPARATION

SF(J,I):FEED RATE OF COMPONENT I AT THE J TH STAGE.

C: TOTAL NUMBER OF COMPONENT I AT THE J TH STAGE.

O: ENTHALPY BALNCE NORMALISATION FACTOR
                NAPHTHALI SANDHOLM METHOD
               LISTING OF SYMBOLS ARE AS
SUBROUTINE ENL :CDMPUTES LIQUID PHASE ENTHALPY OF ALGE
SUBROUTINE ENV :CDMPONENTS FOR A GIVEN STAGE

CDMPONENTS AT A GIVEN STAGE

CDMPONENTS AT A GIVEN STAGE

COMPONENTS AT GIVEN STAGE

COMPONENTS THE DERIVATIVE OF EQUILIBRIUM

PHASE ENTHALPY OF ALL COMPONENTS FOR A

PARTICULAR STAGE

COMPUTES THE DERIVATIVE OF THE VAPOR PHASE

STAGE

STAGE

COMPUTES THE INVERSION OF B SUBMATRIX

NEW VECTORS

NEW VECTORS

INTEGER C.CT

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AREA1/SLO, SVB, TB, TO, VB/AREA2/SS, SSS, TF, ETA, SQ

COMMON / AI/SA/A2/SC/A3/SE

DIMENSION B(29,29), C1(29,29), PB(14), PC(14), C2(8,29,29)
```

```
1,F(29),F1(8,29),A(29,29),SVB(14),SA(14,4),SE(14,4)
2,SL(8,14),SV(8,14),AL(8),V(8),SS(8),SSS(8),
3SF(8,14),SH(14),H(14),SO(8),SHHK(14),HF(14),TF(8),
4ETA(8),AK(14),T(8),HH(14),PA(14),SLO(14),
5VD(20),SC(14,4)
      OPEN (UNIT=21, FILE='S.DAT')

HELLE TERMS MAINLY GUESS VECTOR AND DATA SPECIFICATION
    INPUT TERMS MAINLY GUESS VECTOR A
READ(41,*)N,CT,C
READ(41,*)(SL(J,I),I=1,C),J=1,N)
READ(41,*)(V(J,I),I=1,C),J=1,N)
READ(41,*)(V(J),J=1,N)
READ(41,*)(T(J),J=1,N)
READ(41,*)(T(J),J=1,N)
READ(41,*)(T(J),J=1,N)
READ(41,*)(SO(J),J=1,N)
READ(41,*)(SO(J),J=1,N)
READ(41,*)(SSS(J),J=1,N)
     READ(41,*)TB,TO,VB
READ(41,*)(($A(1,J),J=1,4),I=1,CD
READ(41,*)(($C(I,J),J=1,4),I=1,CD
READ(41,*)(($E(I,J),J=1,4),I=1,C)
READ(41,*)((SE(1,J),J=1,4),I=1,C)

END OF INPUT SPECIFICATION

AEF=0.

IF(IN.GE.1)GOTO 701

WRITE(44,2210)

FORMAT(4,INITIAL GUESS OF FLOW RATES AND TEMPERATURE ARE:')

WRITE(44,2211)

FORMAT(4,2211)

FORMAT(4,2212)

WRITE(44,2213)

FORMAT(4,2213)

FORMAT(4,2213)

FORMAT(4,2213)

WRITE(44,2213)

FORMAT(4,2213)

WRITE(44,2213)

FORMAT(4,2213)

WRITE(44,2213)

WRITE(44,2214)

WRITE(44,2214)
```

```
FORMAT(//,'INITIAL VAPOR AND LIQUID RATE AND TEMPERATURE GUESS

WRITE(44,4467)
WRITE(44,7979)
WRITE(44,13)(J,V(J),AL(J),T(J),J=1,N)
WRITE(44,13)(J,V(J),AL(J),T(J),J=1,N)
FORMAT(//,20x,'CDMPONENT',6x,'TDP,LTQ.FEED',10x,'BDTTDM;VAP.FEE:

10.')
WRITE(44,6101)
FORMAT(20x,10('-'),4x,18('-'),4x,18('-'))
WRITE(44,6102)(I,SLO(I),SVB(I),I=1,C)
FORMAT(24x,I3,7x,2(E18.8,4x))//
FORMAT(24x,I3,7x,2(E18.8,4x))//
FORMAT(27x,'1984')
WRITE(5,14)
WRITE(5,14)
WRITE(5,14)
WRITE(5,14)
WRITE(5,14)
WRITE(5,14)
WRITE(5,15)
FORMAT(/,4x,'DEVELOPED AND PROGRAMMED BY L. FIENK')

** COMPUTATION OF AUGMENTED ERROR FUNCTION AT THE GUESS YECTOR

** # CONTRIBUTION DUE TO MASS BALANCE * #

EVALUATION OF LO(I) & V(N+1,I) FOR TOTAL CONDENSER
  EVALUATION OF LO(I) & V(N+1,I) FOR TOTAL CONDENSER TOTAL REBOILER
END OF EVALUATION OF MASS BALANCE CONTRIBUTION START OF EVALUATION OF THE CONTRIBUTION DUE TO:
```

```
E=0 2 J=1,N-1

TT=T(J)

p1=1.+SS(J)

p2=1.+SS(J)

TFF=TF(J)

IF(J,NE,I) GOTO 601

CALL ENV(TT,SHH)

CALL ENV(TT,SHH)

CALL ENV(TZ,SHH)

IF(J,NE,I) GOTO 26

E==+SLO(I,I)*SH(I)+SV(2,I)*HH(I)+SF(1,I)*HF(I)

SHH(I)=SH(I)

SHH(I)=SH(I)

CONTINUE

F1(I,CT)=-O*F(I,CT)*F1(1,CT)

F1(I,CT)=-O*F(I,CT)*F1(1,CT)

F1(I,CT)=-O*F(I,CT)*F1(1,CT)

F1(I,CT)=-O*F(I,CT)*F1(1,CT)

F1(I,CT)=-O*F(I,CT)*F1(1,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-O*F(I,CT)*F1(I,CT)

F1(I,CT)=-(E+SO(J))

AFP=AEF+F1(J,CT)*F1(J,CT)

F1(I,CT)=-(E+SO(J))

AFP=AEF+F1(I,CT)*F1(J,CT)

F1(I,CT)=-(E+SO(I))

AFP=AEF+F1(I,CT)*F1(J,CT)

F1(I,CT)=-(E+SO(I))

AFP=AEF+F1(I,CT)*F1(I,CT)

F1(I,CT)=-(E+SO(I))

F1(I
               CALL ENV(TB, HH)

CALL ENV(TB, HH)

COD 272 I=1, C

E=E+SL(N-1, I) *SHH(I) +SVB(I) *HH(I) +SF(N, I) *HF(I) -P2*SL(N, I) *SH(I)

1-P1*SV(N, I) *H(I)

CONTINUE

F1(N, CT) =-(E+SO(N))

AEF=AEF+F1(N, CT) *F1(N, CT)

F1(N, CT) = Q*F1(N, CT)

FFF=AEF

WRITE(44, 22)N, AEF

WRITE(5, 22)N, AEF

WRITE(5, 22)N, AEF

**

COMPUTE THE LIQUID & VAPOR FLOW RATE FOR ALL STAGES

**

CONTRIBUTION DUE TO EQUILIBRIUM CONDITION CONSIDERING

EFFICIENCY OF ALL STAGES

**

CONTRIBUTION DUE TO EQUILIBRIUM CONDITION CONSIDERING

EFFICIENCY OF ALL STAGES
```

```
B(CT, CT) = B(CT, CT) *Q

D0 42 I = 1 C

B(I, I) = -6 2

B(I, 
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          ERROR VECTOR*,/(4X,9(E13.6,3X)))
        DO 54 J=2,N
P1=1.+$S(J)
P2=1.+$SS(J)
TT=T(J=1)
CALL DENL(TT,H)
AA=0.
CALL ENL(TT,SHH)
DO 55 I=1,C
AA=AA+SL(J=1,I)*H(I)
A(CT,I)=SHH(I)*Q
A(II,I)=1.
CONTINUE
A(CT,CT)=AA*Q
```

```
END OF COMPUTATION OF ELEMENTS OF A WRITTEN STATE O
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                             ELEMENTS OF A MATRIX
IJ=1,CT),I=1,CT0
OF A MATRIX,/(4x,9(E11,4,3x)))
                                                                                                                                                                                                                                                                                                                                                                                                                                  JID, JI=1, CT), I=4, CT)
                                                                                                                                                                                                                                                                                                                                                                                                                                                  HI, CIT) *P3 * SL(J), I)
```

```
DO 72 JJ=1,N-1
J=N-JJ
DO 75 II (J,CT
F((),T=1,CT
F((),T=1,CT)
C1(I,T)=C2(J,I,II)
C1(I,T)=C2(J,I,II)
C1(I,T)=C2(J,I,II)
C2(I,T)=C2(J,I,II)
C38 M5,305)(F(I),I=1,CT)
C4 M5,305)(F(I),I=1,CT)
C5 M7 M5,305(C1,I),I=1,CT)
C1(I,I)=F(I)+F1(J,I)
C1(J,I)=F(I)+F1(J,I)
C1(J,I)=F(I)+F1(J,I)
C1(J,I)=SV(J,I)-DF*F1(J,I)
C1(J,I)=SV(J,I)-DF*F1(J,I)
C1(J,I)=III
WRITE(I,I)-DF*F1(J,CT)
C1(I,II)=0
C1(I,II)
OF FLOW RATES AND TEMPERATURE ARE: 1
                                                                                                                                                                                            ,12X,'2",12X,'3",12X,'4",12X,'5",12X,'6",12X,
```

```
VAPOR AND LIQUID RATE AND TEMPERATURE ARE: 1//)
                           LIQUID
```

1SA(I,2)+2.\*SA(I,3)\*I+3.\*SA(I,4)\*I\*2)\*(SA(I,1)+SA(I,2)\*I 2+SA(I,3)\*T\*2+SA(I,4)\*T\*3)\*2 CONTINUE RETURN END

## A 91885

Thesis 660.2842 Date Sup 91885  LS4C  This book is to be returned on the date last stamped.	
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